



Standard Test Method for Analysis of *p*-Xylene by Gas Chromatography¹

This standard is issued under the fixed designation D 3798; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope*

1.1 This test method covers the determination of known hydrocarbon impurities in, and the purity of *p*-xylene by gas chromatography (GC). It is generally meant for the analysis of *p*-xylene of 99 % or greater purity. Impurity concentrations that can be measured range from 0.001 to 1.000 weight %.

1.2 The following applies to all specified limits in this test method: for purposes of determining conformance with this test method, an observed value or a calculated value shall be rounded off 'to the nearest unit' in the right-hand digit used in expressing the specification limit, in accordance with the rounding-off method of Practice E 29.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* For a specific hazard statement, see Section 9.

2. Referenced Documents

2.1 ASTM Standards:

- D 3437 Practice for Sampling and Handling Liquid Cyclic Products²
- D 4790 Terminology of Aromatic Hydrocarbons and Related Chemicals²
- D 5136 Specification for High Purity *p*-Xylene²
- E 29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications³
- E 260 Practice for Packed Column Gas Chromatography³
- E 355 Practice for Gas Chromatography Terms and Relationships³
- E 691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method³

¹ This test method is under the jurisdiction of ASTM Committee D16 on Aromatic Hydrocarbons and Related Chemicals and is the direct responsibility of Subcommittee D16.01 on Benzene, Toluene, Xylenes, Cyclohexane and Their Derivatives.

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² *Annual Book of ASTM Standards*, Vol 06.04.

³ *Annual Book of ASTM Standards*, Vol 14.02.

E 1510 Practice for Installing Fused Silica Open Tubular Capillary Columns in Gas Chromatographs⁴

2.2 Other Document:

OSHA Regulations, 29 CFR, paragraphs 1910.1000 and 1910.1200⁵

3. Terminology

3.1 Definitions:

3.1.1 For definition of terms used in this test method see Terminology D 4790.

4. Summary of Test Method

4.1 A known amount of an internal standard is added to a specimen of *p*-xylene. The prepared specimen is mixed and analyzed by a gas chromatograph equipped with a flame ionization detector (FID). The peak area of each impurity and the internal standard is measured. The amount of each impurity is calculated from the ratio of the peak area of the internal standard versus the peak area of the impurity. Purity by GC (the *p*-xylene content) is calculated by subtracting the sum of the impurities found from 100.00. Results are reported in weight percent.

5. Significance and Use

5.1 This test method is suitable for setting specifications on *p*-xylene and for use as an internal quality control tool where *p*-xylene is produced or is used in a manufacturing process. It may also be used in development or research work involving *p*-xylene. It is generally applied to determining those commonly occurring impurities such as nonaromatic hydrocarbons, benzene, toluene, ethylbenzene, *m*-xylene, *o*-xylene, and cumene (isopropylbenzene).

5.2 Purity is commonly reported by subtracting the determined expected impurities from 100.00. However, a gas chromatographic analysis cannot determine absolute purity if unknown components are contained within the material being examined. Refer to Specification D 5136 for determining other chemical and physical properties of *p*-xylene.

⁴ *Annual Book of ASTM Standards*, Vol 03.06.

⁵ Available from Superintendent of Documents, U.S. Government Printing Office, Washington, DC 20402.

*A Summary of Changes section appears at the end of this standard.

6. Interferences

6.1 The internal standard chosen must be sufficiently resolved from any impurity and the *p*-xylene peak. Refer to 8.4.1.

7. Apparatus

7.1 *Gas Chromatograph*—Any chromatograph having a flame ionization detector that can be operated at the conditions given in Table 1. The system should have sufficient sensitivity to obtain a minimum peak height response for a 0.001 weight % impurity twice the height of the signal background noise.

7.2 *Columns*—Different columns have been found satisfactory, depending on the purity of the *p*-xylene to be analyzed.

7.2.1 *p-Xylene Range from 99.0 to 99.8 %*—Both capillary and packed columns have been found satisfactory. The column must give satisfactory separation of the internal standard from *p*-xylene and the impurity peaks. Complete separation of ethylbenzene and *m*-xylene from *p*-xylene is difficult and can be considered adequate if the distance from the baseline to the valley between peaks is not greater than 50 % of the peak height of the impurity. Table 1 contains a description of two columns that have been found satisfactory.

7.2.2 *p-Xylene Range 99.8 % and Greater*—Only capillary columns have been found satisfactory. Complete separation of ethylbenzene and *m*-xylene from *p*-xylene is difficult and can be considered adequate if the distance from the baseline to the valley between peaks is not greater than 50 % of the peak height of the impurity. It is important that tangential skimming is employed. Table 1 contains a description of one column that has been found satisfactory.

7.3 *Recorder*—optional

7.4 Electronic integration is required. Tangent skimming capabilities are required because of the difficulty in fully separating impurities from *p*-xylene.

7.5 *Microsyringe*, 100- μ L capacity.

7.6 *Volumetric Flask*, 100-mL capacity.

8. Reagents

8.1 *Purity of Reagent*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.⁶ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

8.2 *High Purity p-Xylene (99.99 % or greater purity)*—Most *p*-xylene is available commercially at a purity less than 99.9 weight %, but can be purified by recrystallization. To prepare 2 qt of high-purity *p*-xylene, begin with approximately 1 gal of reagent-grade *p*-xylene and cool in an explosion-proof freezer at $-10 \pm 10^\circ\text{C}$ until approximately $\frac{1}{2}$ to $\frac{3}{4}$ of the *p*-xylene has frozen. This should require about 5 h. Remove the sample and decant the liquid portion. Allow the *p*-xylene to thaw and repeat the crystallization step on the remaining sample until the *p*-xylene is free of contamination as indicated by gas chromatography.

8.3 *Carrier Gas*—Chromatographic-grade nitrogen, helium, or hydrogen have been found satisfactory for the *p*-xylene range from 99.0 to 99.8 %. However, only helium or hydrogen have been found satisfactory for the *p*-xylene range of 99.8 % and greater.

8.4 *Pure Compounds for Calibration*, shall include *m*-xylene, *o*-xylene, toluene, ethylbenzene, isopropylbenzene (cumene), and *n*-nonane. The purity of all reagents should be 99 % or greater. If the purity is less than 99 %, the concentration and identification of impurities must be known so that the composition of the standard can be adjusted for the presence of the impurities.

8.4.1 *Internal Standard*—*n*-Undecane (NC11) is the recommended internal standard of choice for Conditions A and *n*-octane (NC8) for Conditions B in Table 1. However, other compounds may be found acceptable provided they meet the criteria as defined in Section 6 and 8.4.

9. Hazards

9.1 Consult current OSHA regulations, supplier's Material Safety Data Sheets, and local regulations for all materials used in this test method.

10. Sampling

10.1 Sample the material in accordance with Practice D 3437.

11. Preparation of Apparatus

11.1 The method used to prepare packed columns is not critical provided that the finished column produces the desired separation.

⁶ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.

TABLE 1 Typical Instrument Conditions for *p*-Xylene Analysis

	A	B
<i>Column:</i>		
Tubing	fused silica	stainless steel
Stationary phase	crosslinked	diisodecylphthalate Bentone 34 ^A
Concentration, weight %	polyethylene glycol	3.5 %/3.5 %
Solid support	not applicable	flux-calcined diatomite ^B
Mesh size	not applicable	60 to 80
Film thickness, μ	0.25	not applicable
Length, m	50	6.1
Inside diameter, mm	0.32	3.2
Carrier gas:	helium	helium
Flow rate, mL/min	1.0	30
Split ratio	100:1	not applicable
<i>Temperature, °C:</i>		
Inlet	200	200
Detector	200	200
Column	60	60
Internal Standard	<i>n</i> -undecane	<i>n</i> -octane

^A The sole source of supply of the material known to the committee at this time is Bentone 34, available from National Lead Co. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

11.2 Follow the manufacturer's instructions for mounting and conditioning the column into the chromatograph and adjusting the instrument to the conditions described in Table 1. Allow sufficient time for the equipment to reach equilibrium. See Practices E 260, E 355, and E 1510 for additional information on gas chromatography practices and terminology.

12. Calibration

12.1 Prepare synthetic mixtures of *p*-xylene with representative impurities on a weight basis. Weigh each hydrocarbon impurity to the nearest 0.0001 g. Refer to Table 2 for an example of a typical calibration blend. *n*-Nonane will represent the nonaromatic fraction. This standard may be purchased if desired.

12.2 Using the exact weight, or alternatively the exact volumes and densities (see Table 2), calculate the weight % concentration for each impurity in each calibration blend of 12.1.

12.3 Into a 100-mL volumetric flask, add 100.0 μL of *n*-undecane to 99.90 mL of the calibration blend; mix well. Assuming a density of 0.861 g/mL for the *p*-xylene blend and 0.740 g/mL for NC11, the resulting NC11 concentration will be 0.086 weight %.

12.4 Inject the resulting solution from 12.3 into the chromatograph. A typical chromatogram is illustrated in Fig. 1.

12.5 Determine the response factor for each impurity relative to NC11 by measuring the area under each peak and calculate as follows:

$$R_i = \frac{(A_s)(C_i)}{(C_s)(A_i)} \quad (1)$$

where:

R_i = response factor for impurity *i* relative to the internal standard,

A_i = peak area of impurity *i*,

A_s = peak area of the internal standard,

C_s = concentration of the internal standard, weight %, and

C_i = concentration of impurity *i*, as calculated in 12.3, weight %.

12.6 Calculate the response factors to the nearest 0.001.

13. Procedure

13.1 Fill 100-mL volumetric flask half full with sample. Pipet 100.0 μL of internal standard into a 100-mL volumetric flask and dilute to the mark with the sample to be analyzed. Mix well.

13.2 Depending upon the actual chromatograph's operating conditions, charge an appropriate amount of specimen into the instrument.

13.3 Measure the area of all peaks except *p*-xylene. Measurements on the specimen must be consistent with those made on the calibration blend. Sum and report the nonaromatic fraction as a total area. A poorly resolved peak, such as *m*-xylene will often require a tangent skim from the neighboring peak. Make consistent measurements on the specimen and calibration chromatograms for tangents or poorly resolved peaks. A typical chromatogram is shown in Fig. 1.

14. Calculation

14.1 Calculate the amounts of each individual impurity. Total the concentration of all impurities. Calculate the *p*-xylene purity by the difference from 100.00.

14.2 Calculate the impurities as follows:

$$C_i = \frac{(A_i)(R_i)(C_s)}{(A_s)} \quad (2)$$

$$C_t = \sum C_i$$

where:

C_t = total concentration of all impurities, weight %.

14.3 Calculate the purity of *p*-xylene, *P*, in weight percent as follows:

$$P = 100.00 - C_t \quad (3)$$

15. Report

15.1 Report the following information:

15.1.1 Individual impurities to the nearest 0.001 weight %,

15.1.2 For concentrations of impurities less than 0.001 weight %, report as <0.001 weight %, and consider as 0.000 in summation of impurities,

15.1.3 The total impurities to the nearest 0.01 weight %, and

15.1.4 *p*-Xylene content to the nearest 0.01 weight %.

16. Precision and Bias ⁷

16.1 The following criteria should be used to judge the acceptability (95 % confidence level) of results obtained by this test method. The criteria for *p*-xylene from 99.0 to 99.8 weight % were derived from a round robin among 13 laboratories. The data were run on 2 days using different operators and three samples ranging in concentration from 99.0 to 99.8 weight %. The criteria for *p*-xylene from 99.8 and greater weight % were derived from a round robin among seven laboratories. The data were run on two days using different operators and two samples ranging in concentration from 99.8 to 99.9 + weight %. Results of the interlaboratory study were calculated using Practice E 691.

16.1.1 *Intermediate Precision, (formerly Repeatability)*—Results in the same laboratory should not be considered suspect unless they differ by more than \pm the amount shown in Table 3 or Table 4 depending on the *p*-xylene concentration. On the basis of test error alone, the difference between two test results obtained in the same laboratory on the same material will be expected to exceed this value only about 5 % of the time.

⁷ Supporting data are available from ASTM International Headquarters. Request RR:D16-1011.

TABLE 2 Typical Calibration Blend

Component	mL	Density ^A	Weight, g	Concentration, weight %
<i>p</i> -Xylene (see 7.2.1)	572.0	0.857	490.2	99.64
Toluene	0.058	0.862	0.050	0.010
Ethylbenzene	0.579	0.863	0.500	0.102
<i>m</i> -Xylene	1.163	0.860	1.000	0.203
<i>o</i> -Xylene	0.116	0.876	0.102	0.021
Isopropylbenzene	0.058	0.857	0.050	0.010
<i>n</i> -Nonane	0.070	0.714	0.050	0.010

^A Numbers are density in grams per millilitre at 25°C.

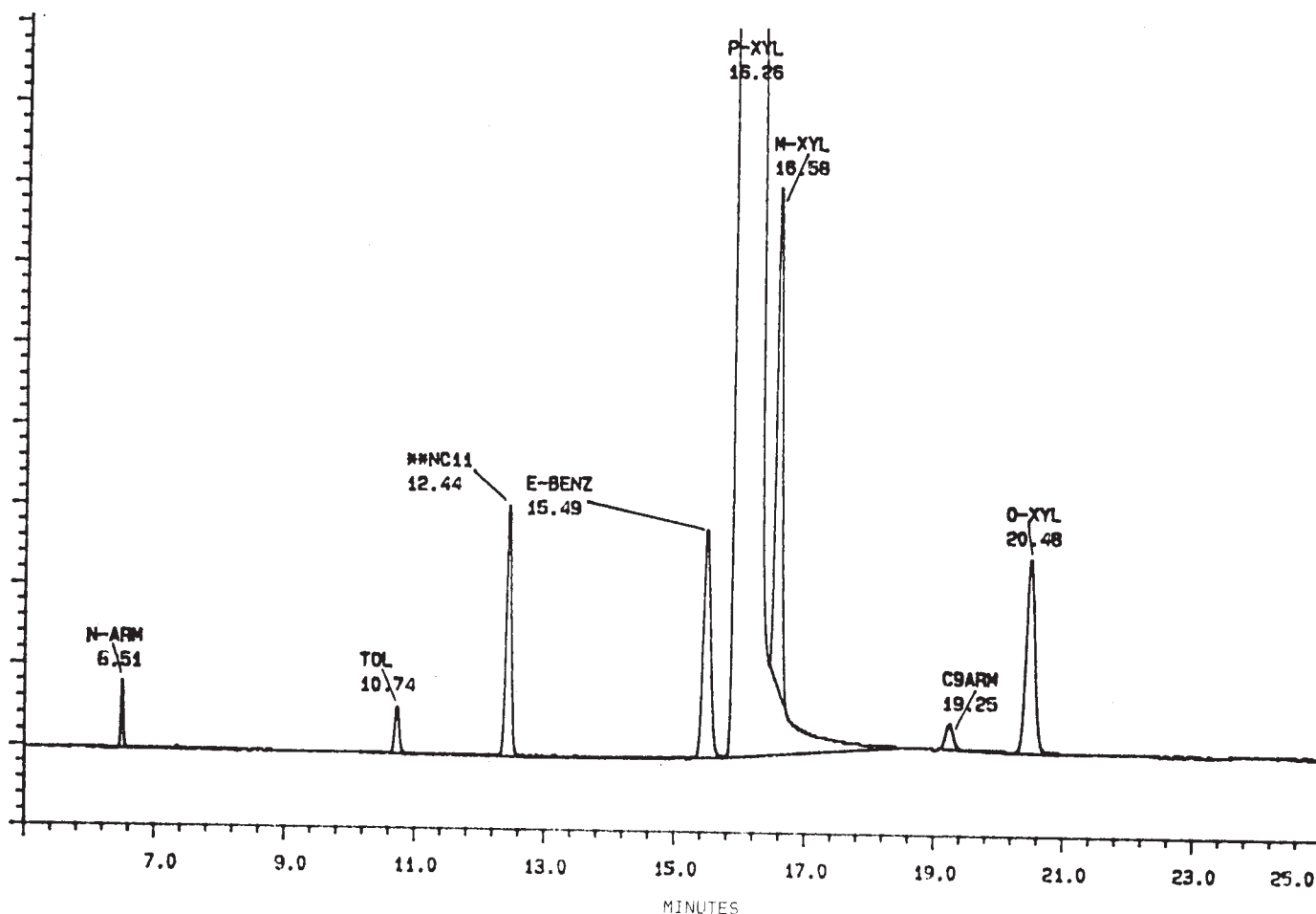


FIG. 1 *p*-Xylene Analysis Using Conditions A in Table 1

TABLE 3 Intermediate Precision and Reproducibility for Internal Standard Where *p*-Xylene Range is from 99.0 to 99.8 %

NOTE 1—This data was calculated after the removal of outliers using Practice E 691. Variation of the *p*-xylene purity was determined from the variation of the calculated total purity.

Component	Concentration Weight %	Intermediate Precision	Reproducibility
<i>p</i> -Xylene	99.510	0.026	0.093
Nonaromatics	0.030	0.014	0.058
Toluene	0.014	0.003	0.009
Ethylbenzene	0.110	0.011	0.029
<i>m</i> -Xylene	0.250	0.018	0.043
<i>o</i> -Xylene	0.100	0.011	0.020

16.1.2 *Reproducibility*—Results submitted by each of two laboratories should not be considered suspect unless they differ by more than the amount shown in Table 3 or Table 4 depending on the *p*-xylene concentration. On the basis of test error alone, the difference between two test results obtained in different laboratories on the same material will be expected to exceed this value only about 5 % of the time.

16.2 *Bias*—The results from the analysis by 13 different laboratories of a gravimetrically prepared blend of *p*-xylene in

TABLE 4 Intermediate Precision and Reproducibility for Internal Standard Where *p*-Xylene Range is 99.8 % or Greater

NOTE 1—This data was calculated after the removal of outliers using Practice E 691. Variation of the *p*-Xylene purity was determined from the variation of the calculated total purity.

Component	Concentration Weight %	Intermediate Precision	Reproducibility
<i>p</i> -Xylene	99.885	0.046	0.055
Nonaromatics	0.007	0.012	0.013
Toluene	0.007	0.003	0.004
Ethylbenzene	0.017	0.006	0.011
<i>m</i> -Xylene	0.061	0.028	0.038
<i>o</i> -Xylene	0.024	0.005	0.010

the concentration range from 99.0 to 99.8 weight % indicates this procedure does not contain a measurable amount of bias nor systematic error that could contribute to a difference between a population mean and the accepted true value. Bias for *p*-xylene in the concentration range of 99.8 and greater weight % has not been determined.

17. Keywords

17.1 1;4-dimethyl benzene; gas chromatography; *p*-Xylene

SUMMARY OF CHANGES

Committee D16 has identified the location of selected changes since the last issue (D 3798 - 00) that may impact the use of this standard.

- (1) Table 1 was modified to read Typical Instrument Conditions for *p*-Xylene Analysis. (2) Section 7.3 was modified to make the recorder optional.
(3) Section 12.1 was modified to allow purchase of standards.

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