



Standard Test Method for Determination of Corrosion and Fouling Tendency of Cooling Water Under Heat Transfer Conditions¹

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 ϵ^1 Note—A footnote was editorially removed in July 1999.

1. Scope

1.1 This test method provides directions for fabricating and operating a test apparatus to simultaneously monitor the corrosion and fouling tendency of real and pilot cooling water systems under heat transfer conditions.

1.2 Interpretation of the results of this test method must be left to the investigator. Many variables are involved which may not be easily controlled or fully understood. Variations in design and operating conditions may produce results that are not comparable from unit to unit.

1.3 The values stated in inch-pound units are to be regarded as the standard. The values given in parentheses are provided for information only.

1.4 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.

2. Referenced Documents

2.1 ASTM Standards:

- D 1129 Terminology Relating to Water²
- D 1193 Specification for Reagent Water²
- D 2777 Practice for Determination of Precision and Bias of Applicable Methods of Committee D-19 on Water²

3. Terminology

3.1 *Definitions*—For definitions of terms used in this test method, refer to Terminology D 1129.

3.2 Definitions of Terms Specific to This Standard:

3.2.1 *corrosion*— the deterioration of the metal by reaction with its environment.

3.2.2 *fouling*— deposition of organic matter or inorganic matter, or both, on heat transfer surfaces that result in the loss of heat transfer efficiency.

² Annual Book of ASTM Standards, Vol 11.01.

3.2.3 heat flux— heat transfer per unit area per unit time.

4. Summary of Test Method

4.1 Water from the system to be tested flows across a heated tube of the desired metallurgy at a constant flow rate and heat flux. Corrosion rate is determined by weight loss while fouling tendency is determined by the deposit weight.

5. Significance and Use

5.1 Deposits on heat transfer surfaces reduce efficiency of the heat exchanger affected. A method for easily determining the corrosion and fouling tendency of a particular water under heat transfer conditions will allow the evaluation of changes in the various system variables such as heat flux, flow velocity, metallurgy, cycles-of-concentration, and treatment schemes on heat exchanger performance.

6. Apparatus (Fig. 1)

6.1 *Test Specimen*— A metal tube of $\frac{3}{8}$ or $\frac{1}{2}$ in. (9.5 or 12.5 mm) outside diameter with sufficient inside diameter to snuggly accommodate the cartridge heater. The tube should be cut to a length sufficient to extend $\frac{1}{2}$ in. (12.5 mm) from each end of the test assembly. If both corrosion and deposition are to be determined, metallurgy of the test specimen should match that of the heat exchanger being modeled.

6.2 *Cartridge Heater*— A $\frac{1}{4}$ or $\frac{3}{8}$ in. (6.2 or 9.5 mm) diameter. Heated surface should be 4 to 8 in. (10 to 20 cm) long with a minimum power rating sufficient to provide 110 % of the heat load required (see Eq 7, 8.2.2). The heater should have an unheated section of sufficient length to allow the center of the heated section to be placed consistently in the center of the test specimen.

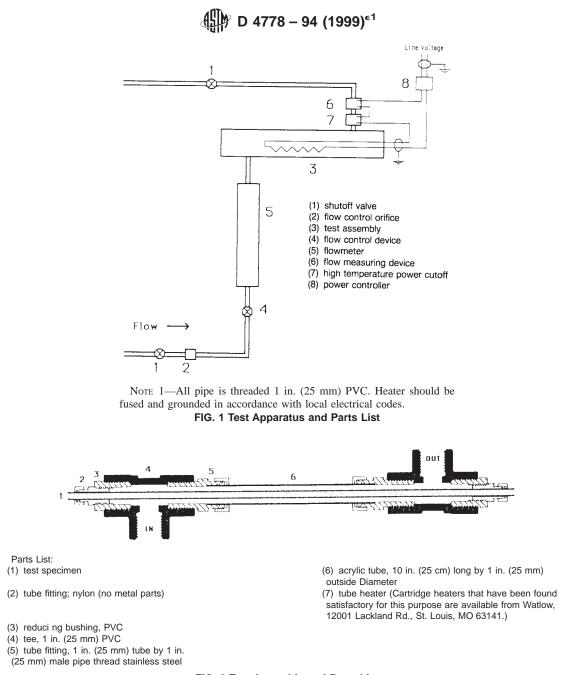
6.3 *Power Controller*— A device to set and control the power to the heater, such as a variable transformer, is used to adjust the heat flux in order to maintain the surface temperature of the test specimen consistent with the heat exchanger being modeled. The power controller should be rated to maintain at least 120 %, but not more than 400 % of the power required.

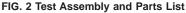
6.4 *Flow Control*— A flow meter or a flow control device such as an orifice, or both, is recommended to maintain a consistent flow rate during the test period.

¹ This test method is under the jurisdiction of ASTM Committee D19 on Water and is the direct responsibility of Subcommittee D19.03 on Sampling of Water and Water-Formed Deposits, Surveillance of Water, and Flow Measurement of Water.

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6.5 *Safety Equipment*— A pressure or flow sensor/controller is necessary to cut power to the heater in the event of a flow interruption. A high temperature cutoff is recommended for added protection.

6.6 Test Assembly— See Fig. 2.

7. Reagents and Materials

7.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society where

such specifications are available.³ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determinations.

7.2 *Purity of Water*— Unless otherwise indicated, references to water shall be understood to mean reagent water as defined by Type III of Specification D 1193.

³ "Reagent Chemicals, American Chemical Society Specifications," Am. Chemical Soc., Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see "Analar Standards for Laboratory Chemicals," BDH Ltd., Poole, Dorset, U. K., and the "United States Pharmacopeia."

7.3 Acetone.

7.4 Hydrochloric Acid (HCl), Inhibited.

- 7.5 Isopropyl Alcohol (C₃H₈O).
- 7.6 Trichloroethylene.

7.7 Trisodium Phosphate —(Na₃PO₄·12H₂O), also available as Na $_{3}PO_{4} \cdot 8H_{2}O$. Either grade is satisfactory.

8. Procedure

8.1 Installation of Test Device:

8.1.1 Placement of the test device with respect to the cooling water system is an important factor in monitoring fouling and corrosion in interpreting the test results. Fouling and corrosion are both affected by temperature. In the case of corrosion, the higher the water temperature, the greater will be the corrosivity of the water. Fouling, however, is a far more complex phenomenon, involving one or more of several types of foulants, namely, particulate matter, precipitates, biomass, corrosion products, and contamination. There are five phases involved in the fouling phenomenon: initiation, attachment, removal, transport, and aging.

8.1.2 Several of the foulants are temperature sensitive. Precipitates, such as calcium carbonate, tend to precipitate more rapidly as temperatures increase. Most biomasses, on the other hand, would agglomerate more rapidly at temperatures between 90 and 105°F (32.2 and 40°C).

8.1.3 The test device may be installed to take its inlet water from one of three locations: cold water supply to a heat exchanger, a heat exchanger outlet, or warm water return to the cooling tower. The choice of location is a function of the type of fouling problem(s) experienced with the particular system. No matter where it is placed, the fouling conditions in the test device should simulate the plant equipment as closely as possible. Specifically, the surface or interface temperature and the shear stress of the water film against the heated surface in the test device should be the same as in the plant equipment being monitored.

8.2 Determination of Setup Conditions:

8.2.1 Calculate plant heat exchange conditions as follows:

$$V_p = 0.408 (F_p) P / (D^2 N)$$
(1)

where:

= water velocity, ft/s, V

= process,

 F^{p} = water flow rate, gal/min,

P = number of passes,

D = number of tubes in process heat exchanger, in., and

N = number of tubes in process heat exchanger.

$$(q/A)_p = 1910 (T_o - T_i) (F_p)/(DLN)$$

where:

= heat flux on inner tube, $Btu/h/ft^2$, q/A

= process,

- T^p = temperature,° F,
- = outlet water, 0

= inlet water,

- F = water flow rate, gal/min,
- D = inside diameter of tube in process heat exchanger, in.,
- L = length of heater section, ft, and
- Ν = number of tubes in process heat exchanger.

$$V_{t} = [V_{p}] [D/(d_{2} - d_{1})]$$
(3)

where:

= water velocity, ft/s, V

= test device, t

process, =

 $\overset{p}{D}$ = inside diameter of tube in process heat exchanger, in.,

 d_2 = inside diameter of outer tube in test device, in., and

= outside diameter of inner (heated) tube in test device, d_1 in.

$$F_t = 2.45 V_t (d_2^2 - d_1^2)$$
(4)

where:

F = water flow rate, gal/min,

= test device,

 V^{t} = water velocity, ft/s,

- d_2 = inside diameter of outer tube in test device, in., and
- d_1 = outside diameter of inner (heated) tube in test device, in.

$$W = 9.8 (T_s - T_b) V^{0.8} L (1 + 0.096 T_b) \text{ (for } d_1 = 0.50)$$
(5)

where:

= power supplied to heater, W, W

Т = temperature, °F,

- = surface or interface, s
- = bulk water, and V^{b}

= water velocity, ft/s,

- L = length of heater section, ft, and
- d_1 = outside diameter of inner (heated) tube in test device, in.

$$W = 7.94 (T_s - T_b) V^{0.8} L(1 + 0.096 T_b) \text{ (for } d_1 = 0.375)$$
(6)

where:

- W = power supplied to heater, W,
- Т = temperature. °F.
- = surface or interface, s
- = bulk water,
- V^{b} = water velocity, ft/s,
- L length of heater section, ft, and =
- d_1 outside diameter of inner (heated) tube in test device, = in.

$$E = (WR)^{0.5}$$

(7)

where:

E = voltage of heater, V,

- = power supplied to heater, W, and W
- R = resistance of heater, ohm.
- 8.3 Preparation of Test Specimen:

(2)

8.3.1 Remove all metal burrs from each end of the tube with a file or emery belt.

8.3.2 Throughly degrease the tube inside and out with a suitable solvent, such as acetone, and brush to remove adherent grease or metal grit.

8.3.3 Dry with a clean cloth and store in a desiccator until dry.

8.3.4 Weigh the clean dry specimen to the nearest milligram.

8.3.5 Store the weighed specimen in a suitable manner (protective atmosphere) to prevent atmospheric corrosion during storage and in transit.

8.4 Assembly of Test Apparatus:

8.4.1 Install earth ground to test apparatus and secure in accordance with local electrical codes.

8.4.2 Remove test specimen from protective atmosphere.

8.4.3 Insert cartridge heater into test specimen to prescribed depth.

NOTE 1-Caution: If the fit is not snug, hot spots may occur and the heater life may be significantly shortened.

8.4.4 Assemble test specimen/cartridge heater into test apparatus using nylon fittings such as Swagelok. Connect heater leads to voltage control device.

8.4.5 Flush inlet water line for 10 min to remove any foreign matter.

8.4.6 Connect inlet and outlet water lines.

8.4.7 Turn water on and adjust flow to that calculated in Eq 4, 8.2.2.

8.4.8 Connect power controller to power source. Turn on power.

8.5 *Operation*:

8.5.1 A minimum test period of 14 days is recommended. A period of 30 to 60 days is preferable in order to more accurately evaluate corrosion and deposition.

8.5.2 Maintain flow and power as constant as possible during the test period, making frequent small adjustments rather than infrequent, but large adjustments when and if fluctuations do occur. Keep a log of all changes and adjustments.

8.6 Analysis:

8.6.1 At the end of the test period, turn off power and disconnect power controller from power source. Then slowly shutdown water flow.

8.6.2 Carefully drain water from test apparatus to prevent disruption of deposit film.

8.6.3 Remove the test specimen from the apparatus without disturbing deposit film. Note the deposit characteristics such as volume, thickness, color, and appearance. Photograph the deposit where possible.

NOTE 2-If there is any delay in transporting the test specimen to the laboratory where the analysis will be performed, then it should be placed in a protective atmosphere in the interim period.

8.6.4 Dry the specimen in a desiccator to constant weight. Weigh to the nearest milligram.

NOTE 3-Deposit may flake off during drying. Place a long sheet of paper under the specimen to collect any fallen deposit and add the weight of this deposit to the test specimen weight.

8.6.5 If the deposit is to be analyzed for composition, remove as much of it as possible with a plastic knife and add to it the deposit collected in 8.6.4.

8.6.6 Clean the test specimen as well as possible with a plastic knife. Remove oily deposits by soaking the specimen in acetone or trichloroethylene. Remove remaining loose deposits from the specimen by wiping with a soft cloth or bristle brush. If the test specimen is clean, proceed to 8.6.7. If adherent deposits remain, continue with one of the following chemical cleaning procedures:

8.6.6.1 For ferrous, copper, and copper alloy specimens, immerse the specimen in inhibited HCl for 30 s at room temperature. Rinse with water after removing the specimen from the inhibited acid bath. Rub the specimen with granular Na_3PO_4 . Rinse with water. Rinse with C_3H_8O . Dry with paper towels followed by warm air drying.

8.6.6.2 Subject a weighed blank coupon of the same metallurgy to the identical cleaning procedure used for the test specimen and reweigh to determine the blank correction factor to be applied to the weight losses.

8.7 After drying, reweigh a clean tube to the nearest milligram.

9. Calculation

9.1 Calculate the deposit weight by subtracting the weight of cleaned test specimen from the weight of specimen with deposit as follows:

$$W_d = W_2 - W_1 \tag{8}$$

where:

 W_d = weight gain due to deposition, mg,

 W_2 = weight of test specimen with deposit, mg, and

 W_I = initial weight of test specimen, mg.

9.2 Calculate the metal weight loss by subtracting the weight of the cleaned test specimen from the initial specimen weight and correcting for the change in weight of a blank after cleaning as follows:

$$W_c = W_1 \left[1 - (W_4 - W_5)/W_4 \right] - W_2$$
(9)

where:

 W_c = weight loss due to corrosion, mg, W_I = initial weight of test specimen, mg,

 W_{\perp} = initial weight of blank specimen, mg,

 W_5 = weight of cleaned blank specimen, mg, and

 W_2 = weight of test specimen with deposit, mg.

9.3 Calculate the average corrosion rate for the test specimen as follows:

$$X_c = (7.09 W_c) / (d_1 L_e Z t)$$

(10)

where:

= average corrosion rate, mills/yr, X_c

 W_c = weight loss due to corrosion, mg,

 d_1 = outside diameter of test specimen, in.,

- L_e = total length of exposed test specimen, in.,
- exposure time, days, and t =

Ζ = density of metal (see Table 1), g/cm^{3} .

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9.4 Calculate the average rate of fouling for the test specimen as follows:

$$X_d = (0.0493 \ W_d) / (d_1 L_h t) \tag{11}$$

where:

- X_d = average rate of fouling, mg/cm²/day,
- $\tilde{W_d}$ = weight gain due to deposition, mg,
- d_i = outside diameter of test specimen, in., L_h = length of heated section, in., and
- = exposure time, days. t

TABLE 1 Density of Metal

Metallurgy	Z			
Admiralty brass	8.52			
Copper	8.94			
Carbon steel	7.86			
304 stainless steel	7.94			

10. Precision and Bias

10.1 Round-robin testing is impractical due to uncontrollable variations in source water that would be encountered at different test sites. Precision and bias of this test method depends on the care taken during removal of the deposit as well as the nature and uniformity of the deposit itself.

10.2 An example of test data obtainable with this test method is shown in Table 2.

10.3 This precision and bias statement meets the requirements of Practice D 2777 and has the approval of the Technical Operations Subcommittee of Committee D-19.

11. Keywords

11.1 cooling water; corrosion; deposits; fouling; heat transfer

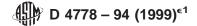


TABLE 2 Test Data

NOTE 1—All tests are at equal heat flux NOTE 2—Holding time index is 11 to 15 h. NOTE 3—Test length is 7 days. NOTE 4—MS = mild steel, ADM = Admiralty brass, and

SS = stainless steel.

Test Number	Unit Number	Treatment	Ca as $CaCO_3$	рН	Notes	Mild Steel MPY	Admiralty MPY	Deposit Weight (mg)		
								MS	ADM	SS
5151	2	1	450	7		1.9	0.1	160	8	11
5156	2	2	450	7		1.6	0	191	86	17
5183	2	3	450	7		1.7	0	97	3	4
5185	2	3	450	7		1.2	0	84	7	8
5142	2	4	450	7		2.7	0	143	21	35
5184	1	4	450	7	Fe = 2 ppm	1.9	0.07	237	2	6
5182	1	4	450	7	Fe = 2 ppm	2.1	0.02	286	20	8
5163	1	5	450	7		2.3	0	55	5	11
5192	1	5	450	7		2.2	0	74	5	7
5190	3	4	600	8		4.4	0	868	38	73
5196	3	4	600	8		5.1	0.04	844	24	35

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